## PREPARATION OF 2,3,6,3',4'-PENTA-Q-ACETYL SUCROSE, THE PRECURSOR OF SUCRALOSE, BY ENZYMATIC METHODS

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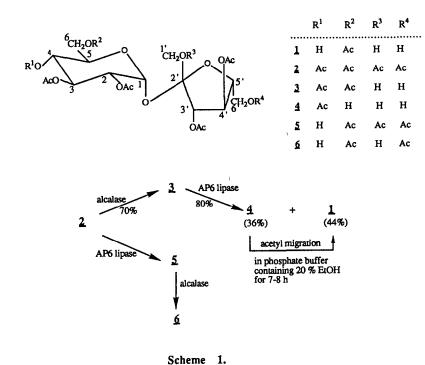
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**Summary**: 2,3,6,3',4'-penta-<u>O</u>-acetyl sucrose, the precursor of sucralose, can be prepared from the sequential hydrolysis of sucrose octaacetate by alcalase and AP-6 lipase.

In 1976, Hough and Phadnis reported the synthesis of 4chloro-4-deoxy-α-D-galactopyranosyl 1,6-dichloro-1,6-dideoxy-β-D-fructofuranoside, called sucralose, and found it to possess sweetness several hundred times that of sucrose1,2. Because of non-nutrition, non-carcinogen and resistance to be hydrolyzed by  $\alpha$ -galactosidase and  $\beta$ -fructofuranosidase, the compound is quite safe for human use. The compound had been synthesized from 2,3,6,3',4'-pentaacetyl sucrose (1) by replacing hydroxyl groups with chlorine, especially the inversion of chirality at C-4 (gluco ---> galacto)3. Compound 1 was obtained from sucrose which was tritylated, acetylated, then detritylated and C4 ---> C6 acetyl migration4,5, but the whole procedure was tedious and inefficient. A strategy that has more applicability for the production of 1 is to selectively remove the acetyl groups from sucrose octaacetate (2). Selective removal of acetyl groups from 2 for the preparation of partially acetylated sucrose has been studied in chemical 6-13 and enzymatic 14-16 ways. Enzymatic methods

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are generally more selective than chemical methods. Recently, a British group has been developed a enzymatic method for preparing 1 directly by the subtilisin-catalyzed hydrolysis of 2<sup>17</sup>. However, the yield was extremely low (about 1.6 %) and quite impractical. Here, we report that 2 prepared from the sequential hydrolysis of 1 by alcalase and lipase AP-6 with 55% of a total yield was shown in Scheme 1.



In order to prepare 1 efficiently and practically, compound 2 was first hydrolyzed by alcalase (subtilisin Carlsberg from Novo, Denmark) to produce 2,3,4,6,3',4'-hexa-Q-acetyl sucrose (3) and then 3 was further hydrolyzed by AP-6 lipase (Aspergillus

nigar from Amano, Japan) to form 2,3,4,3',4'-penta-Q-acetyl sucrose (4) and 1. Compound 4 could be converted to 1 by C<sub>4</sub> ---> C<sub>6</sub> acetyl migration in the phosphate buffer (0.2 M, pH 7.0) containing 20 % ethanol. The structure of 1 and 4 were determined by <sup>1</sup>H-<sup>1</sup>H and <sup>1</sup>H-<sup>13</sup>C 2D-COSY NMR spectra<sup>18</sup>. However, if compound 2 was first hydrolyzed by AP-6 lipase to produce 2,3,6,1',3',4',6'-hepta-Q-acetyl sucrose (5) and then further hydrolyzed by alcalase, the major product was found to be 2,3,6,3',4',6'-hexa-Q-acetyl sucrose (6) with little sucrose penta-acetate. It evidently showed that sucrose hexaacetates were not good substrates for subtilisin. That is reason why 2 could not be obtained directly from the subtilisin-catalyzed hydrolysis of 1 with substantial yield<sup>17</sup>. The method we developed has much higher yield and more practical than that of the British group.

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  N.m.r. data of 4 (CDCl<sub>3</sub>):  $^{1}$ H,  $\delta$  1.97-2.15 (m, 15 H, 5 Ac), 3.18 (t, 1 H, OH-6'), 3.37-3.80 (m, 6 H, H-1',1',6,6,6',6'), 3.95-4.00 (m, 1 H, H-5'), 4.07-4.14 (m, 1 H, H-5), 4.76-4.83 (dd, 1 H,  $\underline{J}_{1,2}$  3.7,  $\underline{J}_{2,3}$  10.3 Hz, H-2), 4.85-4.96 (dd, 1 H,  $\underline{J}_{3,4}$  =  $\underline{J}_{4,5}$  = 7.8 Hz, H-4), 5.34-5.47 (m, 3 H, H-3,3',4'), 5.64 (d, 1 H, H-1);  $^{13}$ C,  $\delta$  20.71-20.81 (5 C, -COCH<sub>3</sub>), 62.89 (C<sub>6</sub>), 63.19 (C<sub>11</sub>), 63.99 (C<sub>6</sub>), 68.89 (C<sub>4</sub>), 70.30 (C<sub>2</sub>), 70.84 (C<sub>5</sub>), 71.96 (C<sub>3</sub>), 74.90 (C<sub>4</sub>), 76.50 (C<sub>31</sub>), 78.80 (C<sub>5</sub>), 90.15 (C<sub>1</sub>), 105.34 (C<sub>2</sub>), 170.21-171.56 (5 C, -COCH<sub>3</sub>).